Chapter 43 Matter as a Wave (Examples) (SM05)

Example 1: What is the wavelength of a 10 eV electron?

Example 2: What is the wavelength of a 45.1 m/s (100 mph) fastball? (mass = 0.14 kg)

Example 3: Find the minimum energy of the electron if an electron microscope is to have a resolution of 5 nm. (Assume the electron's wavelength equals the resolution.)

Example 4: For a hydrogen atom's electron $\Delta x < 0.106$ nm. What is uncertainty in the electron's momentum?

Example 5: A ball of mass 5 gram has a speed of $30.0 \text{ m/s} \pm 0.1 \text{ m/s}$. What is the minimum uncertainty in its location?

Example 6: What is the minimum uncertainty in the energy of an electron trapped inside a nucleus? ($\Delta x < 1.20 \times 10^{-15}$ m)